Concerning the Intermediacy of Uro'gen III and of a Heptacarboxylic Uro'gen in Corrinoid Biosynthesis [*J. Am. Chem. Soc.*, **97**, 2548 (1975)]. By A. IAN SCOTT,* N. GEORGOPAPADAKOU, K. S. HO, S. KLIOZE, E. LEE, S. L. LEE, G. H. TEMME, III, C. A. TOWNSEND, Sterling Chemistry Laboratory, Yale University, New Haven, Connecticut 06520, and I. M. ARMITAGE, Section of Physical Science, Yale School of Medicine, New Haven, Connecticut 06520.

The list of authors is incomplete and should be as follows: A. I. Scott*, B. Yagen, N. Georgopapadakou, K. S. Ho, S. Klioze, E. Lee, S. L. Lee, G. H. Temme, III, C. A. Townsend, and I. M. Armitage. (B. Yagen is presently at the Hebrew University of Jerusalem, The School of Pharmacy, Jerusalem, Israel.)

An Investigation of Potassium Perchromate as a Source of Singlet Oxygen [*J. Am. Chem. Soc.*, **97**, 3299 (1975)]. By JOHN W. PETERS, PAUL J. BEKOWIES, ARTHUR M. WINER, and JAMES N. PITTS, JR.,* Department of Chemistry, University of California, Riverside, California 92502.

The sentence which begins on the last line of the second column on page 3304 (i.e., the second line in the Experimental Section) should read: "A solution of 16.8 g (0.087 mol) of potassium chromate and 3.5 g (0.065 mol) of potassium hydroxide in 200 ml of water was cooled to −5° in an ice–salt bath until crystals began to form." Thus, the words . . . "of potassium chromate and 3.5 g (0.065 mol)" . . . have been omitted.

Molecular Orbital Theory of the Electronic Structure of Organic Compounds. XXIV. Geometries and Energies of Small Boron Compounds. Comparisons with Carbocations [*J. Am. Chem. Soc.*, **97**, 3402 (1975)]. By J. D. DILL, P. v. R. SCHLEYER,* and J. A. POPLE,* Departments of Chemistry, Princeton University, Princeton, N.J. 08540, and Carnegie-Mellon University, Pittsburgh, Pa. 15213.

The structure of eclipsed H_3BOH_2 was inadvertently omitted from Table II. It is described as D–H (H = o,p), C_s symmetry, with parameters BX = 1.624, BH_f = 1.162, BH_g = 1.163, XH = 0.982, H_gBH_h = 114.6, HXH = 105.6, XBH_f = 104.1, XBH_gh = 114.8, BXH_{op} = 129.8.

Photoelectron Spectra of Open Chain C_6H_6 Isomers [*J. Am. Chem. Soc.*, **97**, 5467 (1975)]. By PETER BISCHOF, ROLF GLEITER,* HENNIG HOPF, and FRANK T. LENICH, Institut für Organische Chemie der Technischen Hochschule Darmstadt, D-61 Darmstadt, Germany, and the Institut für Organische Chemie der Universität Karlsruhe, D-75 Karlsruhe, Germany.

Delete the first two lines of the second last paragraph on page 5468 and substitute: "Thus, an estimated basis orbital energy of −10.0 eV for ϵ (π_z) is obtained. This value is in pleasing agreement with . . .".

Melochinone, a Novel Quinolinone from *Melochia tomentosa* L. [*J. Am. Chem. Soc.*, **97**, 6814 (1975)]. By GOVIND J. KAPADIA,* BUDDHA D. PAUL, J. V. SILVERTON, HENRY M. FALES, and EDWARD A. SOKOLOSKI, Department of Biomedical Chemistry, College of Pharmacy and Pharmacal Sciences, Howard University, Washington, D.C. 20059, and the Laboratory of Chemistry, National Heart and Lung Institute, National Institutes of Health, Bethesda, Maryland 20014.

On page 6815, lines 11, 12, and 14, the ions should read *m/e* 332, 331, and 332 rather than 322, 321, and 322 as stated.

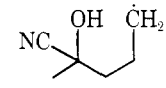
A Method for Determining the Conformational Preference of Nonrigid Radicals by Electron Paramagnetic Resonance [*J. Am. Chem. Soc.*, **98**, 227 (1976)]. By JOHN B. LISLE, LAWRENCE F. WILLIAMS, and DAVID ELDON WOOD,* Department of Chemistry, University of Connecticut, Storrs, Connecticut 06268.

A set of parentheses was left out of eq 2. The correct form is:

$$V(X) = p_1((X/p_2 - 1)(X/p_2 + 1))^2 + (p_3X)^6$$

A Reiterative Functionalization of Unactivated Carbon–Hydrogen Bonds. Photolysis of α-Peracetoxynitriles [*J. Am. Chem. Soc.*, **98**, 271 (1976)]. By DAVID S. WATT, Department of Chemistry, University of Colorado, Boulder, Colorado 80302.

On page 272, structure **8** should be correctly represented as:



8

Dehydration of a Carbinolhydrazine in the Solid State. Correlation with Crystal Structure of the Dehydration of 2-Hydroxy-2-(β-benzoyl-β-phenylhydrazyl)indan-1,3-dione [*J. Am. Chem. Soc.*, **98**, 787 (1976)]. By S. A. PUCKETT, I. C. PAUL,* and D.Y. CURTIN,* Department of Chemistry, University of Illinois, Urbana, Illinois 61801.

In the formula at the lower right-hand corner of Figure 4, there should be a C=N double bond between the disubstituted nitrogen and the indandione ring to which it is attached.

The Hofmann–Loeffler–Freytag Bridge between Mass Spectrometry and Free Radical Chemistry [*J. Am. Chem. Soc.*, **98**, 849 (1976)]. By MARK M. GREEN,* J. M. MOLDOWAN, MARK W. ARMSTRONG, TIM L. THOMPSON, KEVIN J. SPRAGUE, ALAN J. HASS, and JUAN J. ARTUS, Department of Chemistry, Michigan State University, East Lansing, Michigan 48824.

On page 851 in reference 17 (third line), there appears: "and find k_H/k_D is .2 ± 0.2 at 90 °C . . ."; this should read "and find k_H/k_D is 3.2 ± 0.2 at 90 °C . . .".